

## Electrodeposited $(\text{Cu}_2\text{O})_{1-x}(\text{CuO})_x/\text{Cu}_2\text{O}$ heterojunction solar cell

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### Abstract

Cuprous oxide ( $\text{Cu}_2\text{O}$ ) thin films were potentiostatically electrodeposited on Ti substrates in an acetate bath and resulting films were annealed at  $400^\circ\text{C}$  for 15 min in air for the growth of  $p\text{-}(\text{Cu}_2\text{O})_{1-x}(\text{CuO})_x$  composite thin films on Ti. Zero bias spectral response measurements were employed to investigate  $n\text{-Cu}_2\text{O}$  growth conditions on  $p\text{-}(\text{Cu}_2\text{O})_{1-x}(\text{CuO})_x$  composite thin films.  $(\text{Cu}_2\text{O})_{1-x}(\text{CuO})_x/\text{Cu}_2\text{O}$  heterojunction was fabricated for the application of a solar cell by electrodeposition of  $n\text{-Cu}_2\text{O}$  thin film on  $\text{Ti}/(\text{Cu}_2\text{O})_{1-x}(\text{CuO})_x$  electrode at  $-500\text{ mV}$  vs SCE for 60 min. Spectral response, dark and light current-voltage and capacitance-voltage measurements in a PEC were employed in this investigation for the characterization of the thin films.  $\text{Ti}/(\text{Cu}_2\text{O})_{1-x}(\text{CuO})_x/\text{Cu}_2\text{O}/\text{Au}$  solar cell structure was fabricated by sputtering a very thin Au grid on the  $\text{Cu}_2\text{O}$  film and it produced  $V_{\text{OC}}$  of 320 mV and  $I_{\text{SC}}$  of 110  $\mu\text{A}$ .

### 1.0 Introduction

Solar cells are among the most advanced technologies that generate power from abundant solar energy. Silicon is used as the semiconductor material for conventional solar cells but Si solar cells have achieved their maximum efficiencies and they are still very expensive. Therefore scientific community is focusing their attention on low cost thin film semiconductor materials for PV applications. As an alternative material for Si, metal oxide semiconductors play a major role for the development of low cost solar cells. Copper oxides which have two stable forms, cuprous oxide ( $\text{Cu}_2\text{O}$ ) and cupric oxide ( $\text{CuO}$ ), are candidate materials for PV application due to their important optoelectronic properties [1, 2].  $\text{Cu}_2\text{O}$  has a cubic structure with a lattice constant of  $4.27\text{ \AA}$  and a direct optical band gap energy of  $2.0\text{ eV}$ , which is suitable as a window material for solar cells [3, 4].  $\text{Cu}_2\text{O}$  is a defect type semiconductor and it is well understood as a p-type material due to copper vacancies created in the lattice. However, in 1986, it has been reported that electrodeposition of  $\text{Cu}_2\text{O}$  in slightly acidic solutions produces n-type conductivity due to the excess of Cu ions or oxygen vacancies in the  $\text{Cu}_2\text{O}$  lattice [5]. On the other hand,  $\text{CuO}$ , the other stable form of copper oxide, is a p-type semiconductor having a band gap energy of  $1.2\text{ eV}$  which is suitable as an absorber material for PV applications [1].  $\text{Cu}_2\text{O}$  thin films have been prepared by various methods and electrodeposition is an attractive technique because of its advantages such as low cost, simplicity, low temperature process and capability of growth of large area thin films. [2-9]. Growth of  $\text{CuO}$  thin films by annealing of  $\text{Cu}_2\text{O}$  at  $550^\circ\text{C}$  for 30 min in air have been reported earlier and use of these films for application in  $\text{CuO}/\text{Cu}_2\text{O}$  heterojunction solar cell has already been demonstrated [10]. Jayathilaka et al have recently reported the possibility of fabrication of  $\text{CuO}/\text{Cu}_2\text{O}$  heterojunction solar cells by electrodeposition technique [11]. On the other hand, it is very important to explore the possibility of growth of  $(\text{Cu}_2\text{O})_{1-x}(\text{CuO})_x$  composite thin films for possible