

Development of Environmentally Friendly Cu₂O based Low Cost Solar Cell Device



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ABSTRACT

Cuprous oxide (Cu_2O) based solar cell is an attractive photovoltaic device due to its direct band gap of 2 eV, material abundance, non-toxicity and low-cost fabrication. The best approach to improve cell efficiency in Cu_2O based photovoltaic devices is the fabrication of p-n homojunction solar cells. Although electrodeposited p-n homojunction Cu_2O (metal substrate/p- Cu_2O /n- Cu_2O) solar cells were reported earlier, n-p homojunction Cu_2O (metal substrate/n- Cu_2O /p- Cu_2O) solar cells are very limited in the literature. This solar cell structure is very important when exploring the possibilities to improve efficiencies of Cu_2O homojunction solar cells. In the study, Cu_2O thin films were electrodeposited in both acetate and lactate baths. It was observed that by controlling the pH and cupric ion concentration in the deposition bath, both the acceptor (Cu vacancies) and donor levels (O vacancies) can be controlled and thereby it is possible to grow either n-type or p-type Cu_2O .

Most of the reported n-type Cu_2O thin film electrodes exhibited p-type photoactivity in photoelectrochemical cells other than n-type photoactivity and this p-type behavior causes the reduction of overall performance of the cell. This unwanted p-type behavior can be removed by controlling the pH of the deposition bath or by annealing the thin film electrodes in air. The best conditions for the growth of n-type Cu_2O thin film electrodes are potentiostatical electrodeposition at -200 mV in acetate bath pH 6.1 and annealing the electrodeposited n- Cu_2O in air at 160 °C for 30 min or 100 °C for 24 hours. Other than these techniques, growth of a very thin Cu film prior to the deposition of Cu_2O film improved the n-type photosignal. In lactate bath pH around 6.8, p-type photoresponse was observed with 4 M lactic acid, 0.3 M CuSO_4 and 4 M NaOH aqueous solutions. But the best performing p- Cu_2O thin films was achieved by controlling the cupric ion concentration in acetate bath to 0.001 M.

p-n homojunctions of Cu_2O were successfully fabricated by two step electrodeposition of n- Cu_2O followed by p- Cu_2O using acetate and lactate baths. Bias dependence and wavelength dependence of photocurrent of the thin film electrodes in photoelectrochemical cell demonstrated the formation of the p-n homojunction. Annealing the n- Cu_2O /p- Cu_2O homojunction films in air at 175 °C for 20 minutes improves the photoactivity of the films further. The Ti/n- Cu_2O (acetate bath)/p- Cu_2O (acetate bath) homojunction gave the best photoresponse values with an open circuit voltage of -586 mV and a short circuit current of -535 μA in PEC.

Cu_2O n-p homojunction solar cell was fabricated by building front ohmic contacts to the p- Cu_2O layer using partially sulphidation with $(\text{NH}_4)_2\text{S}$ and Na_2S . Direct exposing of $(\text{NH}_4)_2\text{S}$ gas for 8 seconds on p- Cu_2O thin film followed by Na_2S spraying; significantly improved the photocurrent and photovoltage of the cell by reducing the resistance. Sputtering of Au contacts for 2 minutes on Ti/n- Cu_2O /p- Cu_2O /Sulphided Cu_2O cell gives the highest efficiency of 0.89%, V_{oc} of -287 mV and I_{sc} of -12.4 mA/cm^2 under AM 1.5 artificial illumination. Highest reported short circuit current density of 12.4 mA/cm^2 was produced by the Ti/n- Cu_2O /p- Cu_2O /sulphided Cu_2O /Au solar cell structure. The study shows the possibility of fabricating a low cost environmentally friendly n- Cu_2O /p- Cu_2O homojunction for solar cell applications.